FORMATION OF A BIS(TETRAHYDROPYRROLO[1,2-c]-7-PYRIMIDIYL)METHANE UNDER THE CONDITIONS OF THE TROFIMOV REACTION AND IN THE REDUCTION OF 7-FORMYLPYRROLO[1,2-c]PYRIMIDINE

T. N. Borisova, A. É. Aliev, E. A. Sakhnova,

A. A. Sinitsyna, and A. V. Varlamov

We have studied the reduction of 1,2,3,4-tetrahydro-2,4,5-trimethyl-7-formylpyrrolo[1,2-c]pyrimidine (I) with sodium borohydride, as well as with sodium hydroxide in isopropyl alcohol. Instead of the expected methylolderivative, bis(1,2,3,4-tetrahydro-2,4,5-trimethylpyrrolo[1,2-c]-7-pyrimidinyl)methane(II) was obtained in 80% and 60% yields, respectively, in the form of white crystals with mp 140-142°C (dec). The mass spectrum contained a low-intensity peak of a molecular ion with m/z 340 (0.5%); the maximally intense peak of the ion with m/z 176 (100%) is due to cleavage of the $C_{(7)}$ -CH₂ bond and migration of hydrogen from the methylene group to the split-out tetrahydropyrrolopyrimidine fragment. The formation of a secondary alcohol, which is reduced to II, evidently occurs initially.

It was established by chromatographic mass spectroscopy that II is formed as a side product in the synthesis of 1,2,3,4-tetrahydro-2,4,5-trimethylpyrrolo[1,2-c]pyrimidine (III) from 1,3,5-trimethyl-4-piperidone oxime and acetylene in a superbase medium [1]. Compound II was isolated in 0.1% yield.

Me Me
$$C_2H_2$$
 Me H_2O Me H_2O Me H_2O Me H_2O Me H_2O Me H_2O Me H_2O H

Russian International-Friendship University, Moscow 117923. Translated from Khimiya Geterotsiklicheskikh Soedinenii, No. 1, pp. 137-139, January, 1993. Original article submitted December 8, 1992.

It is possible that, under the conditions of the Trofimov reaction, III, which is a cyclic aminal, undergoes cleavage to give formaldehyde. The latter on condensation with two molecules of III is converted to bis(tetrahydropyrrolo[1,2-c]-7-pyrimidinyl)methane II. Similar condensations in the syntheses of di(indolyl)methanes have been studied in detail [2, 3]. PMR spectrum of II (CDCl₃): 1.27 (6H, d, 4-CH₃, $J_{CH3,4} = 6.2$ Hz), 2.02 (6H, s, 5-CH₃), 2.41 (6H, s, 2-CH₃), 2.53 (4H, dd, 3-H_a, $J_{3a,3e} = -12.1$ Hz, $J_{3a,4a} = 6.1$ Hz), 2.74 (4H, dd, 3-H_e, $J_{3e,4a} = 5.4$ Hz), 3.08 (2H, m, 4-H_a), 3.68 (2H, s, CH₂), 4.13 (2H, d, 1-H_e, $J_{1a,1e} = -9.2$ Hz), 4.43 (2H, d, 1-H_a), 5.60 ppm (2H, s, 6-H). ¹³C NMR spectrum (CDCl₃): 11.47 (4-CH₃), 19.61 (5-CH₃), 23.52 (CH₂), 26.90 [C₍₄₎], 42.27 (2-CH₃), 59.00 [C₍₃₎], 66.48 [C₍₁₎], 108.70 [C₍₆₎], 112.54 [C₍₅₎], 124.97 [C₍₇₎], 126.57 ppm [C_(4a)]. Mass spectrum, m/z (I_{rel}, %): 340 (0.5), 325 (0.5), 297 (0.5), 296 (0.5), 282 (15), 176 (100), 163 (15), 161 (25), 134 (18), 120 (10).

REFERENCES

- 1. N. S. Prostakov, A. V. Varlamov, T. N. Borisova, and N. D. Sergeeva, Khim. Geterotsikl. Soedin., No. 9, 1286 (1987).
- 2. E. Leete and L. Marion, Can. J. Chem., 31, 775 (1953).
- 3. J. Thesing, Chem. Ber. 87, 692 (1954).